

A Synthesis of (+)-9a-Desmethoxymitomycin A via Aziridinyl Radical Cyclization

Frederick E. Ziegler* and Michael Y. Berlin

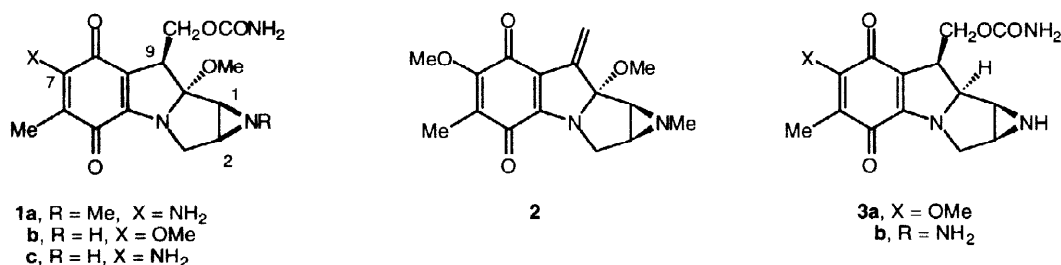
Sterling Chemistry Laboratory

Yale University, New Haven, CT 06511-8118 USA

Received 15 January 1998; accepted 28 January 1998

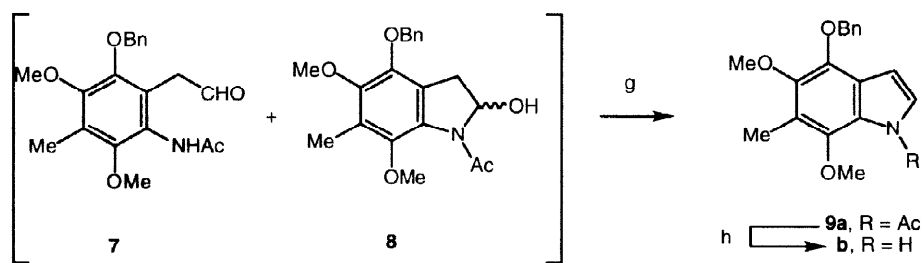
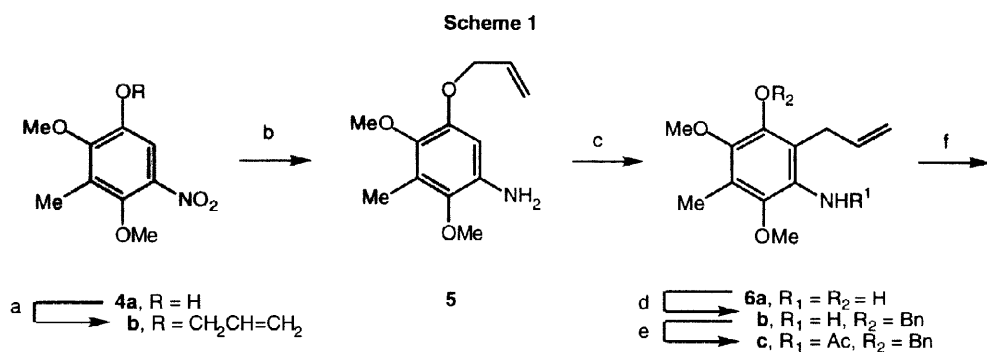
Abstract: A synthesis of (+)-desmethoxymitomycin A is described. The critical step is a highly stereoselective cyclization of an aziridinyl radical to afford a tetracycle having the same relative and absolute stereochemistry as mitomycin A and C. © 1998 Elsevier Science Ltd. All rights reserved.

The antitumor activity¹ and the densely functionalized structure of the mitomycins have made these natural products a target of synthetic efforts over the past twenty years. Notable accomplishments in the field have included the total synthesis of porfiromycin (**1a**),^{2,3} mitomycins A (**1b**)^{4,3} and C (**1c**),^{4,3,5,6} and mitomycin K (**2**).⁷⁻⁹ These syntheses resulted in racemic material; few studies¹⁰ have addressed the synthesis of this class of compounds as their enantiomers. We have developed a protocol for the generation of chiral aziridinyl radicals,¹¹ which we have applied to the synthesis of the nucleus of FR-900482.¹² In this Letter we describe a synthesis of (+)-9a-desmethoxymitomycin A (**3a**) by this general strategy.



m-Nitrophenol **4a** was prepared from 2,6-dimethoxytoluene as previously described by Fukuyama (Scheme 1).¹³ Claisen rearrangement of aniline **5** afforded the hexa-substituted aniline **6a**, which was produced in 51% overall yield from 2,6-dimethoxytoluene. Attempts to effect rearrangement prior to nitro group reduction were unsuccessful. The free-amino group of **6a** required protective acetylation to effect OsO₄/HIO₄ cleavage of the double bond. This process afforded an ~1:1 mixture of uncyclized aldehyde **7** and cyclized alcohol **8**. When the mixture was heated in acetic acid at 80 °C followed by deacetylation with base, indole **9b** was produced in 91% yield from aniline **6a**.

Vilsmeier-Haack formylation of indole **9b** was achieved without incident (Scheme 2). Alkylation of the indole carboxaldehyde **10** was accomplished with the previously described¹¹ chiral aziridinyl triflate **15b** in 78%

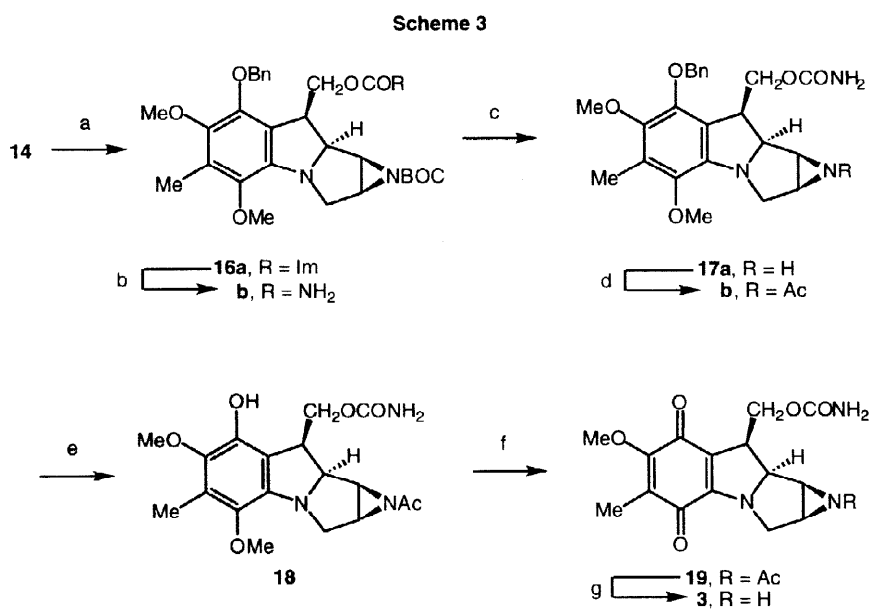
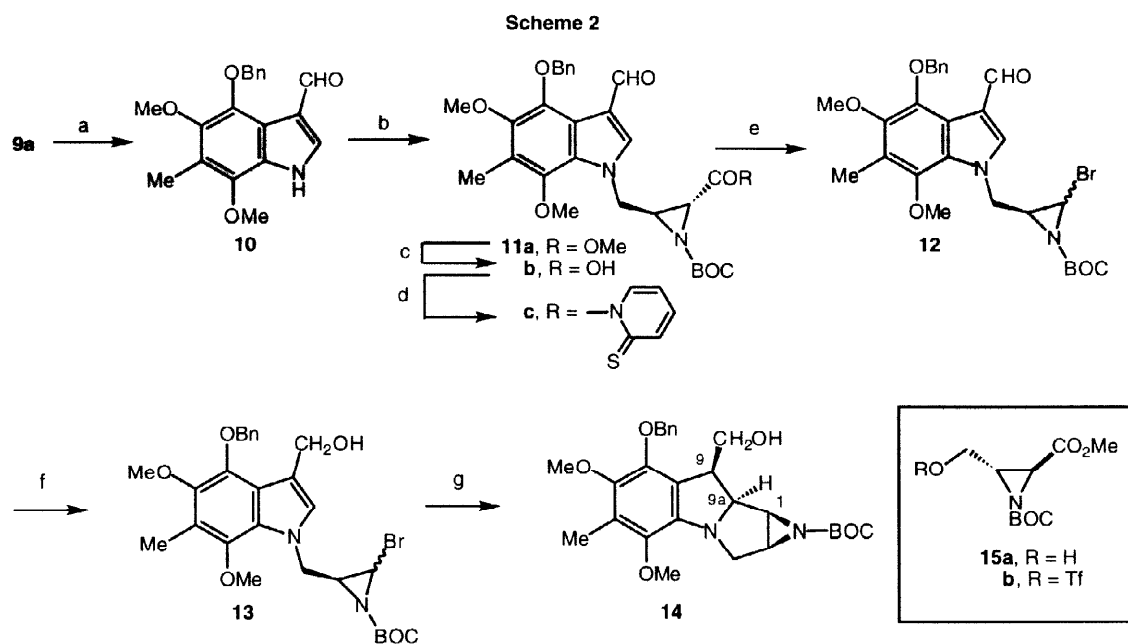


a) CH₂=CHCH₂Br, K₂CO₃, acetone, reflux. b) Zn, HCl, MeOH, 25-50 °C. c) xylenes, 200 °C; 51% from 2,6-dimethoxytoluene. d) BnBr, NaH, Bu₄N⁺I⁻ (cat.), THF, 25 °C. e) Ac₂O, pyr., 25 °C. f) NaIO₄, OsO₄ (cat.), aq. acetone, 25 °C. g) AcOH, 80 °C. h) K₂CO₃, MeOH; 91% from 6a.

yield based on aziridinol **15a**. Activation of the ester group as the thiohydroxamate **11c**¹⁴ and subsequent irradiation in the presence of CBrCl₃ led to a 4:1 mixture of bromoaziridines **12** in 62% yield. The major stereoisomer was assigned to the cis isomer (¹H NMR, *J* = 5.0 Hz) while the minor component was assigned the trans stereochemistry (*J* = 1.7 Hz). The preponderance of the cis isomer presumably reflects the greater steric bulk of the BOC group over the indole moiety in the reaction of the intermediate aziridinyl radical with CBrCl₃. Although the stereoisomers **12** were separable, there was no preparative need to do so. The mixture of bromoaziridines **12** was subjected to reduction with NaBH₄ to provide the 3-indolylmethanols **13**. Subsequent reductive radical cyclization of the mixture **13**, or either of its stereoisomers, in the presence of *n*-Bu₃SnH led to the single tetracyclic alcohol **14** in 55% yield.¹⁵

The relative stereochemistry of alcohol **14** was assigned on the basis of NOE studies (H_{9a} ---> H₁, 10.8%; H_{9a} ---> H₉, 13.3%) and chemical evidence. In an attempt to prepare the triflate of alcohol **14**, intramolecular displacement of the triflate by the BOC group occurred with concomitant loss of the *tert*-butyl group forming the bridged carbamate. This substance can be formed only with the stereochemistry present in **14**. The stereochemistry, relative and absolute, is the same as that found in the mitomycins **1** and it is in accord with previous cyclizations of aziridinyl^{11,12} and oxiranyl^{16,17} radicals.

Introduction of the carbamate moiety at C₁₀ of alcohol **14** was accomplished in a two-step process without isolation of the intermediate imidazolidine **16a** (Scheme 3). Selective removal of the *tert*-butylcarbamate protecting group over the C₁₀ carbamate moiety was accomplished through the use of buffered TMSOTf.¹⁸ Owing to the lability of the unprotected aziridine nitrogen in **17a** toward subsequent oxidation, the nitrogen was reprotected as its



a) $\text{Im}_2\text{C=O}$, THF, 73 °C. b) NH_3 , EtOH-THF, 25 °C; 92% from **14**. c) TMSOTf, 2,6-di-*tert*-butylpyridine, CH_2Cl_2 , 25 °C.
 d) Et_3N , Ac_2O , CH_2Cl_2 , 25 °C; 98% from **16b**. e) H_2 , Pd/C, EtOAc. f) DDQ, aq. acetone, -78 °C \rightarrow 25 °C; 62% from **17b**.
 g) NH_3 (2 equiv.), MeOH; 36%.

acetyl derivative **17b**. Catalytic debenzoylation and DDQ oxidation produced the purple quinone **19** [λ_{\max} (MeOH): 220 (22,200), 319 (12,100), 554 (3,200) nm].

Conversion of mitomycin A (**1b**) to mitomycin C (**1c**)¹⁹ and deacetylation²⁰ of the N-acetylaziridine moiety are standard protocols achieved through ammonolysis in the mitomycin series. Treatment of quinone **19** with a large excess of NH₃/MeOH failed to give rise to 9a-desmethoxymitomycin C (**3b**). However, the use of two equivalents of NH₃ permitted the isolation of the product of deacetylation, (+)-9a-desmethoxymitomycin A [$[\alpha]_{\text{D}}^{23} = +704^{\circ}$ (c, 0.095, MeOH)] in 36% yield. Clearly, the presence of the 9a-methoxy group in mitomycin A affords a stability to that series of compounds that is lacking in the seemingly less complex 9a-desmethoxy series.

Acknowledgments: This research was supported by PHS grant GM-54499.

References and Notes:

- (1) Doyle, T. W.; Bradner, W. T. In *Anticancer Agents Based on Natural Products*; J. M. Cassady and J. D. Douros, Ed.; Academic Press: New York, 1980; Vol. 6; p 43.
- (2) Nakatsubo, F.; Fukuyama, T.; Cocuzza, A. J.; Kishi, Y. *J. Am. Chem. Soc.* **1977**, *99*, 8115.
- (3) Kishi, Y. *J. Nat. Prod.* **1979**, *42*, 549.
- (4) Fukuyama, T.; Nakatsubo, F.; Cocuzza, A. J.; Kishi, Y. *Tetrahedron Lett.* **1977**, 4295.
- (5) Fukuyama, T.; Yang, L. *J. Am. Chem. Soc.* **1987**, *109*, 7881.
- (6) Fukuyama, T.; Yang, L. *J. Am. Chem. Soc.* **1989**, *111*, 8303.
- (7) Benbow, J. W.; Schulte, G. K.; Danishefsky, S. J. *Angew. Chem., Int. Ed. Engl.* **1992**, *31*, 915.
- (8) For an overview of synthetic efforts in the Danishefsky laboratories, see Danishefsky, S. J.; Schkeryantz, J. M. *Syn. Lett.* **1995**, 475.
- (9) Wang, Z.; Jimenez, L. *Tetrahedron Lett.* **1996**, *37*, 6049.
- (10) Shaw, K. J.; Luly, J. R.; Rapoport, H. *J. Org. Chem.* **1985**, *50*, 4515.
- (11) Ziegler, F. E.; Belema, M. *J. Org. Chem.* **1994**, *59*, 7962.
- (12) Ziegler, F. E.; Belema, M. *J. Org. Chem.* **1997**, *62*, 1083.
- (13) Fukuyama, T.; Yang, L. *Tetrahedron Lett.* **1986**, *27*, 6299.
- (14) Barton, D. H. R.; Samadi, M. *Tetrahedron* **1992**, *48*, 7083.
- (15) Efforts to effect cyclizations in the 2-methoxy-3-indolylmethanol series were thwarted by the instability of these intermediates.
- (16) Ziegler, F. E.; Harran, P. G. *Tetrahedron Lett.* **1993**, *34*, 4505.
- (17) Ziegler, F. E.; Wang, Y. *Tetrahedron Lett.* **1996**, *37*, 6299.
- (18) Hamada, Y.; Kato, S.; Shiori, T. *Tetrahedron Lett.* **1985**, *26*, 3223.
- (19) Webb, J. S.; Cosulich, D. B.; Mowat, J. H.; Patrick, J. B.; Broschard, R. W.; Meyer, W. E.; Williams, R. P.; Wolf, C. F.; Fulmor, W.; Pidachs, C.; Lancaster, J. E. *J. Am. Chem. Soc.* **1962**, *84*, 3185.
- (20) Arai, H.; Kasai, M. *J. Org. Chem.* **1994**, *59*, 1087.